Precipitation and the Global Air-Sea CO₂ Flux

Christopher J. Zappa Lamont-Doherty Earth Observatory, Columbia University, NY, USA

Acknowledgments:

W. McGillis – LDEO D. LeBel – LDEO D. Ho – Univeristy of Hawaii D. Turk – Dalhousie University



This work was supported by the Office of Naval Research and the National Science Foundation.

Air-Sea Gas Flux Climatology Workshop

September 26, 2013

Overview

- Komori et al (2007) investigated the effects of impinging rain • and rain deposition on the net global CO2 flux.
- Observations from Biosphere III (Rain) Experiments in 2003
- SSS dilution and dependence on rain rate K_{rain} scales with KEF in saltwater as it does for freshwater
- Describe the effect of Chemical Dilution on the enhancement of • gas exchange.
- As this effect is confined to a very near-surface layer it is neglected in surface mixed layer and climate models as well as by standard measurements of surface pCO_2 that are normally made at 3-5 m depth.
- Quantify the range of precipitation effects on the global CO_2 • flux
- Takahashi et al. (2009) $\Delta p CO_2$ and NCEP II winds
- **Global Precipitation Climatology Program daily rain product**

Global Flux of CO₂



The flux of gas is the product of the gas transfer velocity and the concentration difference between atmosphere and

the ocean



Takahashi et al., 2009

Air-Sea Interaction Processes



Global Flux of CO₂ Due to Rain for 2001







Komori et al., 2007





Impinging raindrops enhance turbulence (Ho et al., [1997, 2000, 2004], Takagaki & Komori [2007], Zappa et al., [2009])



Raindrop comes to equilibrium before impinging (Sugioka and Komori [2007]).



Komori et al. [2007] suggested that the effect of rainfall on the net global CO_2 budget for 2001 is less than 3%. Also found that rainfall effects were significant for the local air-sea CO_2 budget in equatorial and mid-latitude regions compared to the

Biosphere 2 Ocean



- 2,650 m³ of saltwater
- Surface area of 711 m²
- Salinity of 35.5; Water maintained at 26.5 °C
- Vacuum wave generator
- Most of the ceiling above ocean > 13 m

Zappa, C. J., D. T. Ho, W. R. McGillis, M. L. Banner, J. W. H. Dacey, L. F. Bliven, B. Ma, and J. Nystuen (2009), Rain-induced turbulence and air-sea gas transfer, *Journal Of Geophysical Research Oceans, 114(C07009), doi:10.1029/2008JC005008.*



Mean depth of 3.5 m

Bio2 RainX III

• Investigate the effect of freshwater rain on the gas transfer velocity for a saltwater ocean

•Significant dilution of the near-surface salinity in addition to enhancement of turbulence and gas transfer velocity.

Relevant Measurements:

- Gas exchange determined by SF₆ evasion experiment.
- Rain rate and drop size distribution using NASA's Rain Imaging Sensor.
- Monitoring the high-resolution temperature and salinity gradients.
- •RE1: a long rain event (180 min).

•RE2/3: a short rain events (30 min).

•RE4: a longer rain event (122 min), during which an SF₆ evasion experiment was conducted.

Infrared Imagery of Impinging Rain



Effect of Stratification on Near-Surface Turbulence

- Turbulence decays with depth from the air-water interface, trapped by stratification
- Turbulence very near the surface is key to transfer.

Turbulent Dissipation Rate



Stratification



Effect of Stratification on Near-Surface Turbulence

- Turbulence decays with depth from the air-water interface,
- Less Stratified because of background mixing
- Turbulence very near the surface is key to transfer.

Turbulent Dissipation Rate



Stratification



Gas Transfer Scaling



Gas transfer velocity from ACFT versus modeled k as determined from (1) for all rain rates during both RainX II (see *Ho et al.* [2004]) and RainX III at Biosphere 2. The gas transfer measurement using the SF₆ tracer release during RainX III is also shown for comparison.

Mean Turbulence Profile

• Turbulence decays with depth from the air-water interface

• Compared to Craig-Banner model with standard Mellor-Yamada coefficients, the measured KEF input, and variation in zo based on previous field measurements.

• Note: turbulence very near the surface is key to air-sea gas transfer.

Dissipation Rates: Breaking Waves: 10⁻⁵ to 10⁻² W kg⁻¹

Energetic Mixed Layers: 10⁻⁶ to 10⁻⁴ W kg⁻¹

Hudson River: 10⁻⁷ to 10⁻⁵ W kg⁻¹





• Near-surface salinity, total alkalinity (TA), and dissolved inorganic carbon (DIC) are diluted during rainfall in proportion to the excess of precipitation over evaporation. This dilution also decreases $p(CO_2)$ in the surface boundary layer [*Dickson et al.*, 2007].

• Observational data from the model ocean studies [*Zappa et al.*, 2009] showed the rate of sea surface salinity (SSS) dilution is linearly dependent on the measured rain rate.

• Effects of dilution of DIC and TA on $p(CO_2)$ are determined based on standard carbon system models [*Dickson et al.*, 2007].

• Due to the strong stratification, surface effect can be maintained for a significant time.

Field Observations

- Seawater collected from Long Island Sound.
- Realistic chemicals mimicking the ocean surface









Takahashi ΔpCO_2 Database for 2000



The flux of gas is the product of the gas transfer velocity and the concentration difference between atmosphere and



Takahashi et al., 2009

Quantifying the Flux Components



•Transfer velocity due to wind is determined from the quadratic relationship of *Wannhikhof* (1992).



•Transfer velocity for both wind and rain is determined from the sum of the quadratic relationship of *Wannhikhof* (1992) and k_{rain} . The transfer velocity due to rain is determined from Ho et al. (1997) and the results are compared. Ho et al. (2007) have suggested that the k_{wind} and k_{rain} are linearly additive.



• Diluted surface $p(CO_2)$ is ocean surface $p(CO_2)$ after chemical dilution and is determined using a dilution model with measured salinity and rain rate. Strong dilution is typically missed by surface mixed layer and climate models, as well as measurements.

Western Equatorial Pacific



Turk, D., C. J. Zappa, C. S. Meinen, J. R. Christian, D. T. Ho, A. G. Dickson, and W. R. McGillis (2010), Rain impacts on CO_2 exchange in the western equatorial Pacific Ocean, Geophys. Res. Lett., doi:10.1029/24 2010GL045520.

Takahashi ΔpCO₂ Database for 2000



The flux of gas is the product of the gas transfer velocity and the concentration difference between atmosphere and



Takahashi et al., 2009

Global Precipitation Climatology Project

Mean January Rainfall Rate



• Combination of Special Sensor Microwave/Imager (SSM/I) data, infrared (IR) sensor estimates, Atmospheric Infrared Sounder (AIRS data from the NASA Aqua) and Television Infrared Observation Satellite Program (TIROS) Operational Vertical Sounder (TOVS) and Outgoing Longwave Radiation Precipitation Index (OPI) data from the NOAA series satellites..

Global Seasonal Wind Speed





Global Uptake of CO₂ Due to Rain Turbulence



k_{total} from Harrison et al. [2012]

$$k(600)_t = k(600)_w + \left(1 - e^{-a\beta}\right) k(600)_r$$

$$k(600)_r = 63.02(KEF_r)^{0.6242}$$

$$KEF_r = 0.0112R$$



Global Uptake of CO₂ Due to Rain Turbulence and Dilution

k_{total} from Harrison et al. [2012] Raindrops and Dilution



Global Rain Deposition of CO₂



The deposition flux of gas is the product of the rain rate and the concentration of CO_2 in the atmosphere.



Komori et al., 2007

Global Uptake of CO₂ Due to Rain

k_{total} from Harrison et al. [2012] Rain Only Uptake





Global Uptake of CO₂ Due to Wind

Wind Only Uptake



Ratio of Rain to Wind Effects on Global Uptake of CO₂

Rain:Wind Effects



Latitude Variation in the Global Uptake of CO₂ Due to Rain



These two rain processes increase the ocean sink of atmospheric carbon dioxide by more than 0.2 Pg C/year.

Global CO₂ Budget Summary of Results

Takahashi et al. [2009]	Pg C per Year -1.380	Increase in CO ₂ Uptake
Rain-enhanced K without dilution		
Harrison et al. [2012]	-1.379	-0.16 %
Ho et al. [1997]	-1.404	1.61 %
Komori et al. [2007]	-1.385	0.21 %
Rain-enhanced K plus dilution		
Harrison et al. [2012]	-1.453	5.16 %
Ho et al. [1997]	-1.482	7.25 %
Komori et al. [2007]	-1.459	5.61 %
All Rain Effects		
Harrison et al. [2012]	-1.515	9.59 %
Ho et al. [1997]	-1.544	11.69 %
Komori et al. [2007]	-1.521	10.04 %



- Explored the effects of precipitation on the global flux of CO₂, including enhanced transfer velocity, chemical dilution of the near surface, and deposition.
 - Built on the results of Komori et al. (2007) to include the effect of chemical dilution.
- For the year 2000, results show that the effects of precipitation increase the net global CO_2 flux by more than 10%.
 - Applied Harrison et al. [2012] gas transfer wind speed parameterization with enhanced rain effect
 - Compared Ho et al., [1997] and Komori et al., [2007] gas transfer velocity parameterizations for rain
 - Compared assumption for linearly additive k for all wind speeds versus up to a cutoff of 5 m s⁻¹
- Conservative estimate for the flux given the temporal resolution of the global datasets and the nonlinear nature of the enhanced transfer velocity
 - Furthermore, the significant observed stratification may increase the residence time of the surface dilution before it is mixed.
- As ocean uptake is forecast to decrease in time, the contribution from rain may increase to 30% of the total atmosphere to ocean carbon dioxide flux by 2100

Transfer Velocity vs. Wind Speed

• The transfer velocity is a function of the diffusivity and the turbulence at the interface, and is parameterized by $K \propto Sc^{-n} f(u', \ell)$

• Wind forcing plays a central role in generating near-surface turbulence.



courtesy of D. Ho

Relate known K for one gas to any other gas usiKg = K

Gas Transfer with KEF



Gas transfer velocity versus kinetic energy flux, KEF. Freshwater data represent distinct raindrop sizes with diameters of 2.3 mm, 2.8 mm, and 4.2 mm from experiments at the Wallops rain facility and are summarized in *Ho et al.* [2000]. Saltwater data represent broad raindrop size distributions and are from RainX II (see *Ho et al.* [2004]) and this study of RainX III at Biosphere2.



• Equatorial Pacific ΔpCO_2 database from 1990 to 2004 [Christian et al., 2008].

• Examine the decadal variability in the CO₂ uptake by the equatorial Pacific.



Takahashi et al., 2009



•Examine the effects of increasing temporal resolution on globally integrated CO2 uptake

•Deposition expected to be unaffected as it is based on volume of rainfall

•Instantaneous gas transfer rates predicted to be larger

Daily: GPCP daily rainfall climatology, daily NCEP wind speed climatology
Sub-daily: TRMM three-hourly rainfall data (±50 latitude), Merra re-analysis hourly wind speeds

• Preliminary results from the daily climatologies suggest an additional 2.5% increase in the oceanic uptake of CO2

Kwajalein (Marshall Islands)





Figure 32. Rainfall for Kwajalein Island, RMI [NCDC].

- Measure the physical and chemical effects of rain in an open ocean environment ocean.
- Improve our model for the chemical dilution due to rain and the impacts of mixing (e.g., wind and waves).
- Investigate the effects of deposition on the chemistry of the near-surface ocean including the raindrop chemistry.

Average Monthly Rainfall

NEXRAD Base Reflectivity



Interpolated S-band reflectivity at an altitude of 500 meters elevation, in dBZ (40 dBZ = 20 mm/hr rainfall), out to range of 250 km

David E. Weissman Hofstra University

Air-Sea Interaction Processes



DopBeam Measurements Profiles of velocity along the beam









DopBeam Measurements

Profiles of velocity along the beam (vertical)

Frequency Spectrum of Vertical Velocity at single bin



Infrared Imagery and CFT Measurements









Controlled Flux Technique



Frames at 30 Hz

Faster Decay = Faster Renewal = Faster Transfer

Turbulence and Transfer Velocity

Turbulent Dissipation Rate

Transfer Rates from CFT



Gas Transfer Scaling



Gas transfer velocity from ACFT versus modeled k as determined from (1) for all rain rates during both RainX II (see *Ho et al.* [2004]) and RainX III at Biosphere 2. The gas transfer measurement using the SF₆ tracer release during RainX III is also shown for comparison.

Mean Turbulence Profile

• Turbulence decays with depth from the air-water interface

• Compared to Craig-Banner model with standard Mellor-Yamada coefficients, the measured KEF input, and variation in zo based on previous field measurements.

• Note: turbulence very near the surface is key to air-sea gas transfer.

Dissipation Rates: Breaking Waves: 10⁻⁵ to 10⁻² W kg⁻¹

Energetic Mixed Layers: 10⁻⁶ to 10⁻⁴ W kg⁻¹

Hudson River: 10⁻⁷ to 10⁻⁵ W kg⁻¹

